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#### Abstract

3-Acetyl-4-hydroxy-chromen-2-one (1) was brominated with phenyltrimethylammonium tribromide to afford 3-(2-bromoacetyl)-4-hydroxy-chromen-2-one (2) whose reactions with thiourea, thioacetamide and ammonium dithiocarbamate gave respectively 3-(2-amino-thiazol-4-yl)-4-hydroxy-, 4-hydroxy-3-(2-phenyl-thiazol-4-yl)- and 4-hydroxy-3-(2-mercapto-thiazol-4-yl)chromen-2-one. In a similar manner, compound 2 was treated with four 1 -substituted-2-thioureas and thiobenzamide to give the corresponding 4-hydroxy-3-(thiazol-4-yl)-chromen-2-one derivatives.


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In pharmacology, 2-aminothiazoles are one of the most important compounds as precursors to synthetic drugs, such as sulphathiazole (antibiotic) and thiabedazole (antihelmintic). The 2 -aminothiazoles are prepared by Hantzsch reaction of $\alpha$-haloketones with thioureas [1-6]. As an extension of 3-acetyl-4-hydroxy-chromen-2-one chemistry [7-8], now we wish to report the preparation of 3-(2-bromoacetyl)-4-hydroxy-chromen-2-one (2) as new synthon and the synthesis of 4-hydroxy-3-(thiazol-4-yl)-chromen-2-ones by its reactions with thioureas, thioamides and ammonium dithiocarbamate.

Results and Discussion.
Preparation of 3-(2-Bromoacetil)-4-hydroxy-chromen-2one (2).

3-(2-Bromoacetyl)-4-hydroxy-chromen-2-one (2) is expected to be a very useful compound for the synthesis of a various heterocycle-substituted chromen-2-ones (coumarins). However, its preparation by using bromine is difficult, since the chromen-2-one nucleus is very susceptible to electrophilic substitution [9-11]. For example, bromination of 3-acetyl-4-hydroxy-chromen-2-one (1) in a conventional manner (bromine/acetic acid) gave substitution products at the aromatic nucleus as a major product [12-15]. Then, we carried out bromination by using phenyltrimethylammonium tribromide [16-19], which is a useful reagent for $\alpha$-bromination of ketones and acetals.


Figure 1

When 3-acetyl-4-hydroxy-chromen-2-one (1) was treated with phenyltrimethylammonium tribromide in tetrahydrofuran to give only 3-(2-bromoacetyl)-4-hydroxy-chromen-2-one (2) as yellow crystals (mp 144$146^{\circ}$ ). Its structure was determined from the spectral data as well as elemental analysis $\left(\mathrm{C}_{11} \mathrm{H}_{7} \mathrm{BrO}_{4}\right)$. In the ir spectrum, three characteristic absorptions were observed at 3185 (OH), 1725 (bromoacetyl $\mathrm{C}=\mathrm{O}$ ) and $1685 \mathrm{~cm}^{-1}$ (chromen-2-one $\mathrm{C}=\mathrm{O}$ ). The ${ }^{1} \mathrm{H} \mathrm{nmr}$ spectrum shows a singlet peak at $\delta 4.28(2 \mathrm{H})$ for $\mathrm{CH}_{2}$, signals at $\delta$ 7.31-7.67 $(4 \mathrm{H})$ for ring protons and broadened singlet peak at $\delta$ $15.70(\mathrm{OH})$.

Synthesis of 4-Hydroxy-3-(thiazol-4-yl)coumarines.
Treatment of 3-(2-bromoacetyl)-4-hydroxy-chromen-2one (2) with thiourea in refluxing ethanol for 30 minutes gave 3-(2-amino-thiazol-4-yl)-4-hydroxy-chromen-2-one hydrobromide (3a) as yellow needles ( $\mathrm{mp} 255-257^{\circ}$ ) in a $60 \%$ yield. Its structure was determined on the basic of the spectral data as well as on elemental analysis $\left(\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S} \cdot \mathrm{HBr}\right)$. The ir spectrum shows three characteristic absorptions at $3381(\mathrm{OH}), 3120\left(\mathrm{NH}_{2}\right)$ and 1693 $\mathrm{cm}^{-1}(\mathrm{C}=\mathrm{O})$. In the ${ }^{1} \mathrm{H} \mathrm{nmr}$ spectrum, one isolated signal is observed at $\delta 7.46$ (s) for $5^{\prime}-\mathrm{H}$, besides multiplet peaks at $\delta$ 7.29-7.82 (4H), for aromatic protons, broadened singlet peak at $\delta 15.87$ for hydroxyl proton and singlet peak at $\delta$ 8.58 for $\mathrm{NH}_{2}$ protons. $\mathrm{R}_{\mathrm{f}}$ value for this compound is 0.25 using methyl-ethyl ketone:toluene, $1: 9(\mathrm{v} / \mathrm{v})$, as eluent. Compound 3a also gave positive coloration with iron(III) chloride solution.

The reaction of 2 with 1-(methyl)thiourea gave 4-hydroxy-3-(2-methylamino-thiazol-4-yl)-chromen-2-one (3b) $\left(\mathrm{C}_{13} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}\right.$ ), in $67 \%$ yield (mp 218-220 $)$. The methylamino group was confirmed from the ${ }^{1} \mathrm{H} \mathrm{nmr}$ spectrum, which shows a singlet peak at $\delta 2.98(3 \mathrm{H})$ for methyl group and a broad peak at $\delta 11.01(1 \mathrm{H})$ for the NH proton.


Figure 2

In this spectrum, one isolated singlet is observed at $\delta 7.38$ for 5 '- H , besides multiplet peaks at $\delta 7.29-7.84(4 \mathrm{H})$, for aromatic protons and a broadened peak at $\delta 16.36$ for the OH proton. Rf value for this compound is 0.42 (methylethyl ketone:toluene, $1: 9, \mathrm{v} / \mathrm{v}$ )
In a similar manner, treatment of 2 with three 1(aryl)thioureas gave the corresponding 3-(2-aryl-thiazol-4yl )-chromen-2-ones ( $\mathbf{3 c} \mathbf{c}$ ) in 63-74 \% yields. These structures were also determined from the spectral data and elemental analysis (See: Experimental).

When a mixture of 2 and thioacetamide in ethanol was refluxed, 4-hydroxy-3-(2-methyl-thiazol-4-yl)-chromen-2-one ( $\mathbf{4 a}$ ) $\left(\mathrm{C}_{13} \mathrm{H}_{9} \mathrm{NO}_{3} \mathrm{~S}\right)$ was isolated as yellow needles in a $72 \%$ yield (mp 182-184 $)$. Its structure was determined from the spectral data and elemental analysis. In the ir spectrum, two characteristic absorptions were observed at $3378(\mathrm{OH})$, and $1697(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1}$. In the ${ }^{1} \mathrm{H} \mathrm{nmr}$ spectrum, one isolated singlet is observed at $\delta 8.28$ for $5^{\prime}-\mathrm{H}$, besides multiplet peaks at $\delta 7.29-7.80(4 \mathrm{H})$ for aromatic


Figure 3
protons, peak at $\delta 2.46(3 \mathrm{H})$ for $\mathrm{CH}_{3}$ group, and a broad peak at $\delta 15.36$ for OH .

Compound 2 was also reacted with thiobenzamide to afford the corresponding 4-hydroxy-3-(2-phenyl-thiazol4 -yl)-chromen-2-one ( $\mathbf{4} \mathbf{b}$ ), in $74 \%$ yield (mp 189-191 ${ }^{\circ}$ ). The structure of this product was confirmed from its spectral data (See: Experimental).

Furthermore, compound 2 was heated with ammonium dithiocarbamate in refluxing ethanol to give 4-hydroxy-3-(2-mercapto-thiazol-4-yl)-chromen-2-one (5) (mp 204$206^{\circ}$ ) in a $81 \%$ yield. $\mathrm{R}_{\mathrm{f}}$ value for this compound is 0.42 (methyl-ethyl ketone:toluene, 1:9, v/v).


Figure 4

In its ${ }^{1} \mathrm{H}$ nmr spectrum, the SH and $5^{\prime}$ - H protons were observed at $\delta 3.46$ (s) and 8.23 (s), respectively. The elemental analysis $\left(\mathrm{C}_{12} \mathrm{H}_{7} \mathrm{NO}_{3} \mathrm{~S}_{2}\right)$ also supported the structure.
In summary, it is found that 3-(2-bromoacetyl)-4-hydroxy-chromen-2-one (2) is useful material for the synthesis of heterocycle-substituted 4-hydroxy coumarin compounds.

Some of 4-hydroxy-3-(thiazol-4-yl)-chromen-2-one derivarives (as a $N[55$-(4-hydroxy-chromen-2-one)-thia-zol-2'-yl]-benzenesulphnamide [20]) are readily prepared and expected to be pharmaceutical precursors.

## EXPRIMENTAL

Measurements.
Melting points were recorded on a Kofler-hot stage apparatus and are uncorrected. Microanalysis of carbon, hydrogen and nitrogen was carried out with a Carlo Erba 1106 microanalyser. The ir spectra were run on Perkin-Elmer Grating Spectrophotometers Model 137 and Model 197. The nmr spectra were recorded on a VARIAN FT 80 A and 200" Gemini spectrometer, in $\mathrm{CDCl}_{3}$ and DMSO- $\mathrm{d}_{6}$, using TMS as the internal standard. Chemical shifts are given in $\delta(\mathrm{ppm})$; J, coupling constants in hertz (Hz), abbreviations: s-singlet, d-doublet, t-triplet, q-quartet, m-multiplet and br-broadened). Thin layer chromatography was taken on plastic sheets of silica gel 60 (Merck). Abbreviations used: $\mathrm{PhTAPBr}_{3}$-phenyltrimethylammonium tribromide, DMSO-dimethylsulphoxide-d ${ }_{6}$, EtOH-ethanol, $\mathrm{CDCl}_{3}$ deuterochloroform, r.t.-room temperature.

Starting Compound: 3-Acetyl-4-hydroxy-chromen-2-one (1).
To a solution of 4-hydroxy-chromen-2-one ( $3 \mathrm{~g}, 18.6$ mmoles) in acetic acid ( 16 ml ) phosphorous oxychloride ( 5.6 ml ) was added. The mixture was heated at reflux for 30 minutes. After cooling, the precipitate was collected and recrystallized from ethanol, to give 3-acetyl-4-hydroxy-chromen-2-one (1), as white needles, in a yield of $2.7 \mathrm{~g}(90 \%), \mathrm{mp} 134-136^{\circ}$; ir ( KBr ): 3185, $2950,1700,1705,1610,1560,1460.1310,1130,950,840,820$, $770 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}\left(200 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 2.43\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.35$ (ddd, $1 \mathrm{H}, 6-\mathrm{H}^{3} \mathrm{~J}_{6,5}=7.8 \mathrm{~Hz},{ }^{4} \mathrm{~J}_{6,8}=1.2 \mathrm{~Hz},{ }^{3} \mathrm{~J}_{6,7}=7.4 \mathrm{~Hz}$ ), 7.42 (dd, $1 \mathrm{H}, 8-\mathrm{H},{ }^{3} \mathrm{~J}_{7,8}=8.35 \mathrm{~Hz},{ }^{4} \mathrm{~J}_{6,8}=1.2 \mathrm{~Hz}$,), 7.42 (ddd, $1 \mathrm{H}, 7-\mathrm{H}$, $\left.{ }^{3} \mathrm{~J}_{7,8}=8.35 \mathrm{~Hz},{ }^{3} \mathrm{~J}_{7,6}=7.4 \mathrm{~Hz},{ }^{4} \mathrm{~J}_{7,5}=1.6 \mathrm{~Hz}\right), 7.69(\mathrm{dd}, 1 \mathrm{H}, 5-\mathrm{H}$, $\left.{ }^{3} \mathrm{~J}_{5,6}=7.8 \mathrm{~Hz},{ }^{4} \mathrm{~J}_{5,7}=1.6 \mathrm{~Hz}\right), 15.7(\mathrm{OH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}(50 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): \delta 28.33\left(\mathrm{CH}_{3}\right), 160.10(\mathrm{CO}), 177.32(\mathrm{C}-4), 116.91(\mathrm{C}-$ 8), 159.65 (C-2), 154.10 (C-9), 136.85 (C-7), 116.09 (C-5), 124.82 (C-6), 114.41 (C-10), 101.91 (C-3); ms: m/z 204 $\left(\mathrm{M}^{+}, 100\right) 189(74), 161(43), 120(17), 119(31), 92(56), 78(33)$, 43 (28).
Anal. Calcd. for $\mathrm{C}_{11} \mathrm{H}_{8} \mathrm{O}_{4}: \mathrm{C}, 64.71 ; \mathrm{H}, 3.95$. Found: C, 64.92; H, 3.68.

3-(2-Bromoacetyl)-4-hydroxy-chromen-2-one (2).
To a solution of 3-acetyl-4-hydroxy-chromen-2-one (1) (2.0 g, 9.8 mmoles ) in tetrahydrofuran ( 40 ml ) was added phenyltrimethylammonium tribromide ( $3.68 \mathrm{~g}, 9.8$ mmoles) in a period of 15 minutes (at room temperature). A precipitate was deposited from the solution, and the color of the solution changed into pale yellow. After stirring for 20 minutes and standing for 30 minutes, cold water ( 100 ml ) was added to the reaction mixture. The precipitate was collected, washed with water and recrystallized from ethanol to afford 3-(2-bromoacetyl)-4-hydroxy-chromen-2-one (2) as light yellow needles, in yield $2.51 \mathrm{~g}(90 \%)$, mp 144-146 ${ }^{\circ}$; ir (KBr): $3185,1725,1685,1560,1437,1200$, 1032, $945,842,822,771 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(200 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta$ 4.28 (s, $2 \mathrm{H}, \mathrm{CH}_{2}$ ), 7.31 (ddd, $1 \mathrm{H}, 6-\mathrm{H}, \mathrm{J}_{6,5}=7.38 \mathrm{~Hz},{ }^{4} \mathrm{~J}_{6,8}=1.16$ $\mathrm{Hz},{ }^{3} \mathrm{~J}_{6,7}=7.32 \mathrm{~Hz}$ ), $7.39\left(\mathrm{dd}, 1 \mathrm{H}, 8-\mathrm{H},{ }^{3} \mathrm{~J}_{7,8}=8.35 \mathrm{~Hz},{ }^{4} \mathrm{~J}_{6,8}=1.16\right.$ $\mathrm{Hz}), 7.42\left(\mathrm{ddd}, 1 \mathrm{H}, 7-\mathrm{H},{ }^{3} \mathrm{~J}_{7,8}=8.35 \mathrm{~Hz},{ }^{3} \mathrm{~J}_{7,6}=7.37 \mathrm{~Hz}\right.$, ${ }^{4} \mathrm{~J}_{7,5}=1.63 \mathrm{~Hz}$,), $7.67\left(\mathrm{dd}, 1 \mathrm{H}, 5-\mathrm{H},{ }^{3} \mathrm{~J}_{5,6}=7.89 \mathrm{~Hz},{ }^{4} \mathrm{~J}_{5,7}=1.63\right.$ Hz, ), $15.70(\mathrm{OH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}\left(50 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 33.41\left(\mathrm{CH}_{2}\right)$, 183.50 (CO), 185.32 (C-4), 115.91 (C-8), 158.65 (C-2), 153.00 (C-9), 134.85 (C-7), 125.09 (C-5), 124.82 (C-6), 119.41 (C-10), 100.91 (C-3).

Anal. Calcd. for $\mathrm{C}_{11} \mathrm{H}_{7} \mathrm{BrO}_{4}$ : C, $46.67 ; \mathrm{H}, 2.49$. Found: C, 46.92; H, 2.38.

Reactions of 3-(2-Bromoacetyl)-4-hydroxy-chromen-2-one (2) with Thioureas.

To a solution of 3-(2-bromoacetyl)-4-hydroxy-chromen-2-one (2) $(1 \mathrm{~g}, 3.5 \mathrm{mmoles})$ in absolute ethanol $(60 \mathrm{ml})$ thiourea ( 3.5 mmoles) was added. The mixture was refluxed for 30 minutes. After cooling, the precipitate was collected and recrystallized from ethanol-10\% sodium hydroxide, to give 3-(thiazol-4-yl)-4hydroxy coumarines (3a-e).
3-(2-Amino-thiazol-4-yl)-4-hydroxy-chromen-2-one Hydrobromide (3a).
This compound was obtained from the reaction with thiourea as yellow needles (ethanol-10\% sodium carbonate) in a yield of $0.71 \mathrm{~g}(60 \%)$, mp $255-257^{\circ}$; ir (KBr): 3433, 3381, 3241, 3120, 1693, 1609, 1524, 1405, 1328, 1294, 1165, 1072, $950 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ nmr ( 200 MHz, DMSO-d $\mathrm{d}_{6}$ ): $\delta 7.29-7.37$ (m, 2H, $6-\mathrm{H}, 8-\mathrm{H}$,
$\left.{ }^{4} \mathrm{~J}_{6,8}=1.16 \mathrm{~Hz},{ }^{3} \mathrm{~J}_{6,5}=7.90 \mathrm{~Hz},{ }^{3} \mathrm{~J}_{6,7}=7.35 \mathrm{~Hz},{ }^{3} \mathrm{~J}_{8,7}=8.35 \mathrm{~Hz}\right)$, 7.46 (s, 1H, $5^{\prime}-\mathrm{H}$ ), 7.44 (ddd, $1 \mathrm{H}, 7-\mathrm{H},{ }^{3} \mathrm{~J}_{6,7}=7.35 \mathrm{~Hz},{ }^{4} \mathrm{~J}_{7,5}=1.63$ $\mathrm{Hz},{ }^{3} \mathrm{~J}_{8,7}=8.35 \mathrm{~Hz}$,), $7.82\left(\mathrm{dd}, 1 \mathrm{H}, 5-\mathrm{H},{ }^{4} \mathrm{~J}_{7,5}=1.63 \mathrm{~Hz},{ }^{3} \mathrm{~J}_{6,5}=\right.$ 7.90 Hz ,), 8.58 (bs, $1 \mathrm{H}, \mathrm{NH}_{2}$ ), 15.87 (s, $1 \mathrm{H}, \mathrm{OH}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr} \mathrm{( } 50$ MHz, DMSO-d ${ }_{6}$ ): $\delta 165.42$ (C-2'), 140.67 (C-4'), 108.56 (C-5'), 154.28 (C-2), 93.86 (C-3), 163.09 (C-4), 123.76 (C-5), 124.06 (C-6), 132.11 (C-7), 116.34 (C-8), 120.23 (C-9), 152.05 (C-10).

Anal. Calcd. for $\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S} \cdot \mathrm{HBr}$ : C, 55.37 ; $\mathrm{H}, 3.10$; N, 10.76. Found: C, $55.12 ; \mathrm{H}, 2.98 ; \mathrm{N}, 10.38$.

4-Hydroxy-3-(2-methylamino-thiazol-4-yl)-chromen-2-one (3b).
This compound was obtained from the reaction with 1 -(methyl)-thiourea as yellow needles (from ethanol) in a yield of $0.52 \mathrm{~g}(67 \%), \mathrm{mp} 218-220^{\circ}, \mathrm{R}_{\mathrm{f}}=0.42$ (silica gel, methyl-ethyl ketone:toluene, 1:9, v/v); ir (KBr): 3433, 3381, 3155, $3116(\mathrm{OH})$ and (NH), 1698 (C=O), 1630, 1513, 1409, 1340, 1299, 1185, $1092,960 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}_{6}\right): \delta 2.98\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.29-$ $7.84\left(\mathrm{~m}, 4 \mathrm{H}, 6-\mathrm{H}, 7-\mathrm{H}, 5-\mathrm{H}, 8-\mathrm{H},{ }^{4} \mathrm{~J}_{7,5}=1.63 \mathrm{~Hz},{ }^{4} \mathrm{~J}_{6,8}=1.18 \mathrm{~Hz}\right.$, ${ }^{3} \mathrm{~J}_{8,7}=8.35 \mathrm{~Hz},{ }^{3} \mathrm{~J}_{6.5}=7.89 \mathrm{~Hz},{ }^{3} \mathrm{~J}_{6,7}=7.37 \mathrm{~Hz}$,), $7.38\left(\mathrm{~s}, \mathrm{H}, 5^{\prime}-\mathrm{H}\right)$, 11.01 (bs, 1H, NH), 16.36 (s, 1H, OH); ${ }^{13} \mathrm{C} \mathrm{nmr} \mathrm{(DMSO-d}{ }_{6}$ ): $\delta$ 159.42, (C-2'), 109.46, (C-5'), 141.67 (C-4'), 156.32 (C-2), 168.39 (C-4), $32.34\left(\mathrm{CH}_{3}\right)$.

Anal.Calcd.for $\mathrm{C}_{13} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}: \mathrm{C}, 56.92 ; \mathrm{H}, 3.67$; N, 10.26. Found: C, 56.62; H, 3.98; N, 10.28.

4-Hydroxy-3-(2-phenylamino-thiazol-4-yl)-chromen-2-one (3c).
This compound was obtained from the reaction with 1-(phenyl)-thiourea as yellow needles (from ethanol) in a yield of $0.42 \mathrm{~g}(70 \%), \mathrm{mp} 224-226^{\circ}, \mathrm{R}_{\mathrm{f}}=0.64$ (silica gel, methyl-ethyl ketone:toluene, 1:9, v/v); ir (KBr): 3483 (NH), 3261 (OH), 3030, 1684, (C=O), 1615, 1533, 1459, 1165, 1072, $966 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ nmr (DMSO-d ${ }_{6}$ ): $\delta \quad 7.37-7.85(\mathrm{~m}, 4 \mathrm{H}, 6-\mathrm{H}, 7-\mathrm{H}, 5-\mathrm{H}, 8-\mathrm{H}$, ${ }^{4} \mathrm{~J}_{7,5}=1.63 \mathrm{~Hz},{ }^{4} \mathrm{~J}_{6,8}=1.17 \mathrm{~Hz},{ }^{3} \mathrm{~J}_{8,7}=8.34 \mathrm{~Hz},{ }^{3} \mathrm{~J}_{6,5}=7.87 \mathrm{~Hz}$, $\left.{ }^{3} \mathrm{~J}_{6,7}=7.35 \mathrm{~Hz}\right)$ 6.8-7.2 (m, 5H, phenyl), 8,11 ( $\left.\mathrm{s}, 1 \mathrm{H}, 5^{\prime}-\mathrm{H}\right), 10.31$ (bs, $1 \mathrm{H}, \mathrm{NH}$ ), $12.85(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): ~ \delta$ 159.42 (C-2'), 119.36, (C-5'), 150.37 (C-4'), 156.32 (C-2), 168.78 (C-4).

Anal.Calcd.for $\mathrm{C}_{18} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}: \mathrm{C}, 64.27 ; \mathrm{H}, 3.61 ; \mathrm{N}, 8.33$. Found: C, 63.98; H, 3.78; N, 8.28.

4-hydroxy-3-(2- p-tolylamino-thiazol-4-yl)-chromen-2-one (3d).
This compound was obtained from the reaction with 1-(4methylphenyl)thiourea as yellow needles (from ethanol) in a yield of $0.78 \mathrm{~g}(63 \%), \mathrm{mp} 208-210^{\circ}, \quad \mathrm{R}_{\mathrm{f}}=0.58$ (silica gel, methyl-ethyl ketone:toluene, 1:9, v/v); ir (KBr) 3287 ( OH ), 3147 (NH), 3082, 1692 (C=O), 1621, 1591, 1432, 1356, 1106, $1052,866 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H} \mathrm{nmr}\left(\right.$ DMSO-d $\left.{ }_{6}\right): \delta 2.23\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.29-$ $7.82\left(\mathrm{~m}, 4-\mathrm{H}, 6-\mathrm{H}, 7-\mathrm{H}, 5-\mathrm{H}, 8-\mathrm{H},{ }^{4} \mathrm{~J}_{7,5}=1.63 \mathrm{~Hz},{ }^{4} \mathrm{~J}_{6,8}=1.15 \mathrm{~Hz}\right.$, $\left.{ }^{3} \mathrm{~J}_{8,7}=8.37 \mathrm{~Hz},{ }^{3} \mathrm{~J}_{6,5}=7.89 \mathrm{~Hz},{ }^{3} \mathrm{~J}_{6,7}=7.34 \mathrm{~Hz}\right), 7.12-7.21(\mathrm{ABq}$, 4 H , phenyl, ${ }^{3} \mathrm{~J}=8,41 \mathrm{~Hz}$ ), $8,24(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 9.86(\mathrm{bs}, 1 \mathrm{H}, \mathrm{NH})$, $14.22(\mathrm{bs}, 1 \mathrm{H}, \mathrm{OH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{DMSO}_{\mathrm{d} 6}\right): \delta 20.67\left(\mathrm{CH}_{3}\right), 153.92$ (C-2'), 118.56 (C-5'), 150.67 (C-4'), 145.62 (C-1, phenyl), 129.78 (C-4, phenyl).
Anal. Calcd. for $\mathrm{C}_{19} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}: \mathrm{C}, 64.13 ; \mathrm{H}, 4.03$; $\mathrm{N}, 7.99$. Found: C, 63.99; H, 4.10; N, 8.13.
4-Hydroxy-3-[2-(4-methoxy-phenylamino)-thiazol-4-yl]-chromen-2-one (3e).

This compound was obtained from the reaction with 1-(4methoxyphenyl)thiourea as pale orange needles (from ethanol) in a yield of $0.964 \mathrm{~g}(74 \%), \mathrm{mp} 198-200^{\circ}, \mathrm{R}_{\mathrm{f}}=0.46$ (silica gel,
methyl-ethyl ketone:toluene, 1:9, v/v); ir (KBr): 3476 (NH), $3251(\mathrm{OH}), 3076,1686$ (C=O), 1665, 1563, 1459, 1365, 1002, $846 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}_{6}\right): \delta 7.35-7.85$ (m, 4H, 6-C, 7-C, 5C, $8-\mathrm{C},{ }^{4}{ }^{4}{ }_{7,5}=1.61 \mathrm{~Hz},{ }^{4} \mathrm{~J}_{6,8}=1.15 \mathrm{~Hz},{ }^{3} \mathrm{~J}_{8,7}=8.31 \mathrm{~Hz},{ }^{3} \mathrm{~J}_{6,5}=7.85$ $\mathrm{Hz},{ }^{3} \mathrm{~J}_{6,7}=7.33 \mathrm{~Hz}$ ), 6.81-7.05 (ABq, 4H, phenyl, ${ }^{3} \mathrm{~J}=8,90 \mathrm{~Hz}$ ), $8,17$ (s, 1H, 5'-H), 10.09 (bs, $1 \mathrm{H}, \mathrm{NH}), 14.85$ (bs, $1 \mathrm{H}, \mathrm{OH}$ ); ${ }^{13} \mathrm{C}$ nmr (DMSO-d $\mathrm{d}_{6}$ : $\delta 154.52$ (C-2'), 118.86 (C-5'), 150.27 (C-4'). 153.32 (C-4), 139.78 (C-1).

Anal. Calcd. for $\mathrm{C}_{19} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{4} \mathrm{~S}: \mathrm{C}, 62.28 ; \mathrm{H}, 3.85 ; \mathrm{N}, 7.65$. Found: C, 62.56; H, 3.79; N, 7.48.

Reaction of 3-(2-Bromoacetyl)-4-hydroxychromen-2-one (2) with Thioamides.

A mixture of 3-(2-bromoacetyl)-4-hydroxychromen-2-one (2) ( $1 \mathrm{~g}, 3.5$ mmoles) and thioamide ( 3.5 mmoles ) in absolute ethanol ( 90 ml ) was heated for 30-45 min under refluxing. After cooling, the precipitate was collected and recrystallized to give 4-hydroxy-3-(thiazol-4-yl)-chromen-2-one (4a-b).

4-Hydroxy-3-(2-methyl-thiazol-4-yl)-chromen-2-one (4a).
This compound was obtained from the reaction with methylthioamide as orange needles (ethanol) in a yield of 0.186 g ( $72 \%$ ), $\mathrm{mp} 182-184^{\circ}, \mathrm{R}_{\mathrm{f}}=0.32$ (silica gel, methyl-ethyl ketone:toluene, $1: 9, \mathrm{v} / \mathrm{v})$; ir ( KBr ) $3378(\mathrm{OH}), 3030,1697(\mathrm{C}=\mathrm{O}), 1615,1533$, 1459, 1390, 1165, 1072, $966 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta 2.46$ (s, $\left.3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.29-7.80\left(\mathrm{~m}, 4 \mathrm{H}, 6-\mathrm{H}, 7-\mathrm{H}, 5-\mathrm{H}, 8-\mathrm{H},{ }^{4} \mathrm{~J}_{7,5}=1.62 \mathrm{~Hz}\right.$, ${ }^{4} \mathbf{J}_{6,8}=1.15 \mathrm{~Hz},{ }^{3} \mathbf{J}_{8,7}=8.37 \mathrm{~Hz},{ }^{3} \mathrm{~J}_{6,5}=7.9 \mathrm{~Hz},{ }^{3} \mathbf{J}_{6,7}=7.39 \mathrm{~Hz}$ ), 8.28 (bs, $\left.1 \mathrm{H}, 5^{\prime}-\mathrm{H}\right), 15.36$ (bs, $1 \mathrm{H}, \mathrm{OH}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (DMSO-d ${ }_{6}$ ): $\delta$ 162.02 (C-2'), 111.96 (C-5'), 144.27 (C-4'). 154.32 (C-2), 167.99 (C-4), $18.38\left(\mathrm{CH}_{3}\right)$,

Anal. Calcd. for $\mathrm{C}_{13} \mathrm{H}_{9} \mathrm{NO}_{3} \mathrm{~S}: \mathrm{C}, 60.22$; H, 3.50; N, 5.40. Found: C, 59.86; H, 3.82; N, 5.28.
4-Hydroxy-3-(2-phenyl-thiazol-4-yl)-chromen-2-one (4b).
This compound was obtained from the reaction with phenylthioamide as orange needles (from ethanol) in a yield of 0.186 g (74 \%), mp 189-191, $\mathrm{R}_{\mathrm{f}}=0.45$ (silica gel, methyl-ethyl ketone:toluene, 1:9); ir (KBr) $3165(\mathrm{OH}), 3011,1691(\mathrm{C}=\mathrm{O})$, 1619, 1532, 1439, 1373, 1163, 1077, $969 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (DMSO$\left.\mathrm{d}_{6}\right): \delta 7.10-7.82(\mathrm{~m}, 9 \mathrm{H}$, phenyl and $6-\mathrm{H}, 7-\mathrm{H}, 5-\mathrm{H}, 8-\mathrm{H}), 8.29(\mathrm{~s}$, $\left.1 \mathrm{H}, 5^{\prime}-\mathrm{H}\right), 15.19$ (s, 1H, OH); ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta 167.12$ (C$\left.2^{\prime}\right), 112.96$ (C-5'), 144.87 (C-4'), 155.92 (C-2), 168.77 (C-4).

Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{11} \mathrm{NO}_{3} \mathrm{~S}: \mathrm{C}, 67.28 ; \mathrm{H}, 3.45 ; \mathrm{N}, 4.36$. Found: C, 66.92; H, 3.81; N, 4.28.
4-Hydroxy-3-(2-mercapto-thiazol-4-yl)-chromen-2-one (5).
A mixture of 3-(2-bromoacetyl)-4-hydroxy-chromen-2-one (2) ( $0.285 \mathrm{~g}, 1 \mathrm{mmoles}$ ) and ammonium dithiocarbamate $(0.110 \mathrm{~g}, 1$
mmoles) in absolute ethanol ( 30 ml ) was heated for 30 minutes under refluxing. The precipitate was collected and recristallized from $70 \%$ ethanol to afford 4-hydroxy-3-(2-mercapto-thiazol-4-yl)-chromen-2-one (5) as yellow-orange needles, yield 0.39 g ( 81 $\%$ ), mp 204-206, $\mathrm{R}_{\mathrm{f}}=0.42$ (silica gel, methyl-ethyl ketone:toluene, 1:9, v/v); ir (KBr): $3305(\mathrm{OH}), 3021,1682$ (C=O), 1621, 1533, 1474, 1396, 1166, 1067, $962 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (DMSO): $\delta 3.46$ (s, 1H, SH), 8.23 (s, 1H, 5 '-H), 15.23 (s, 1H, OH ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (DMSO): $\delta 153.12$ (C-2'), 112.86 (C-5'), 144.37 (C-4'), 156.22 (C-2), 168.77 (C-4).

Anal. Calcd. for $\mathrm{C}_{12} \mathrm{H}_{7} \mathrm{NO}_{3} \mathrm{~S}_{2}: \mathrm{C}, 51.97 ; \mathrm{H}, 2.54, \mathrm{~N}, 5.05$. Found: C, 51.55; H, 2.91; N, 5.28.

## REFERENCES AND NOTES

[1] A. Hantzsch, Ann. Chem., 249, 1 (1888).
[2] R. H. Wiley, D. C. England and I. C. Behr, Org. React., 6, 367 (1951).
[3] P.G. Sammes, "Sulfonamides and Sulfones, Comprehensive Medicinal Chemistry", Vol. 2, Eds. Pergamon, Oxford 1990. pp 255-270.
[4] S. Rover, M. A. Cesura, P. Huguenin and A. Szente, J. Med. Chem. 40, 4378 (1997).
[5] L. W. Wattenberg, L. K. T. Low and A. V. Fladmoe, Cancer Res., 39, 1651 (1979); R. E. Willette and T. O. Soine. J. Pharm. Sci., 51, 149 (1961).
[6] F. M. Dean, "Naturally Occurring Oxygen Ring Compounds", Butterworth, London, 1963, pp 176-220.
[7] Chang-Yi Qian, Zhong-Tian Jin and Bing-Zhu Yin, J. Heterocyclic Chem., 26, 601 (1989).
[8] I. Iwataki, Bull. Chem. Soc. Japan, 45, 3218 (1972)
[9] T. Nozoe, "Non-benzenoid Aromatic Compounds", D.
Ginsburg, ed, Interscience Publishers, Inc., New York 1959, pp 339-463.
[10] H. J. M. Dou and G. Vernin and J. Metzger, J. Heterocyclic
Chem., 6, 575 (1969).
[11] M. Baule, R. Vivaldi, J. C. Poite, H. J. M. Dou, G. Vernin and J. Metzger, Bull. Soc. Chim. France, 2679 (1972).
[12] Z.-H. Li, Z.-T. Jin B.-Z. Yin and K. Imafuku, J. Hetericyclic Chem., 24, 779 (1971).
[13] K. Takase, K. Kasai, K. Shimizu and T. Nozoe, Bull. Chem. Soc. Japan, 44, 2466 (1971).
[14] V. Grakauskas, J. Org. Chem., 35, 723 (1970).
[15] V. Grakauskas, J. Org. Chem., 34, 3825 (1969).
[16] M. Fieser and L. F. Fieser, "Reagents of Organic Synthesis", Vol 4, 1974, p 386.
[17] G. Rosini and G. Baccolini, J. Org. Chem., 39, 826 (1974).
[18] M. Gaudry and A. Marquet, Org. Synth., 55, 24 (1976).
[19] S. Viswewariah, Synthesis, 309 (1982).
[20] S. Sukdolak, S Solujic, N. Manojlovic, M. Milosev, and Lj. Krstic, $3^{\text {rd }}$ Int. Conf. of the Chemical Societes of the South-Estern European Countries Vol. 2, PO 422 Bucharest 2002.

